

Supporting Information

A Fast Kinetics Study of the Reactions of Transient Silylenes with Alcohols. Direct Detection of Silylene-Alcohol Complexes in Solution.

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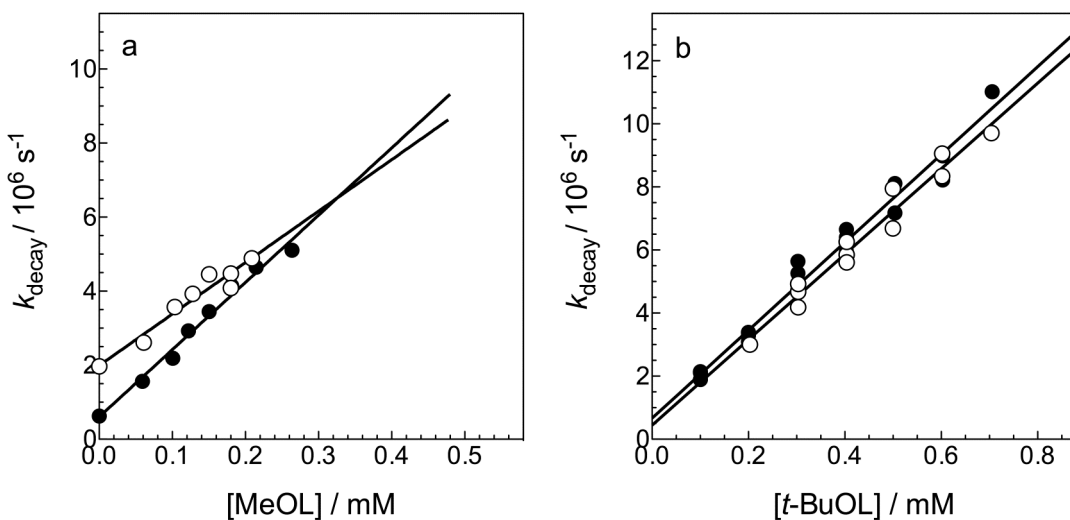


Figure S1. Plots of the pseudo-first order decay rate coefficients (k_{decay}) of SiPh_2 vs. $[\text{ROL}]$ ($L = \text{H}$ (●) or D (○)) for (a) $R = \text{Me}$ and (b) $R = t\text{-Bu}$, in hexanes at $25\text{ }^\circ\text{C}$. The solid lines are the best linear least squares fits of the data to equation 3. The data were obtained in the same experiments as those represented in Figure 3 of the paper.

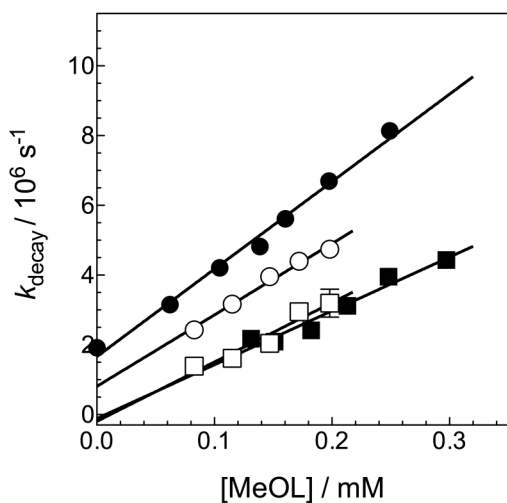


Figure S2. Plots of the pseudo-first order rate coefficients (k_{decay}) for the decay of free SiMe_2 ($\lambda_{\text{max}} = 465\text{ nm}$; $L = \text{H}$ (●) and D (○)), and of the $\text{SiMe}_2\text{-MeOL}$ complex (monitored at 310 nm ; $L = \text{H}$ (■) and D (□)), vs. $[\text{MeOL}]$. The solid lines are the best linear least squares fits of the data to equation 3.

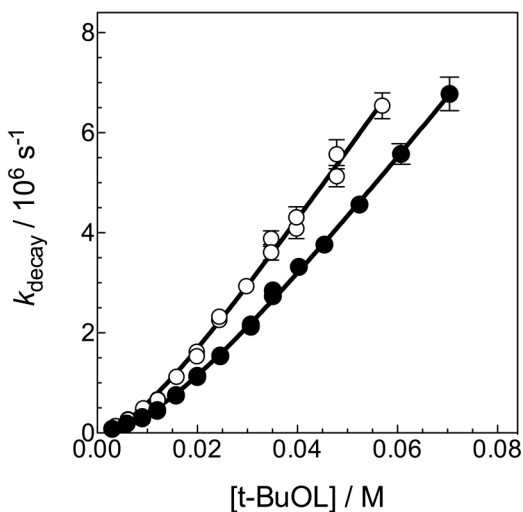


Figure S3. Representative plots of the pseudo-first order rate coefficients for the decay of SiMes_2 vs. $[\text{t-BuOL}]$ ($L = \text{H}$, (●) or D (○)) in hexanes at 25°C . The solid lines are the best non-linear least squares fits of the data to equation 5.

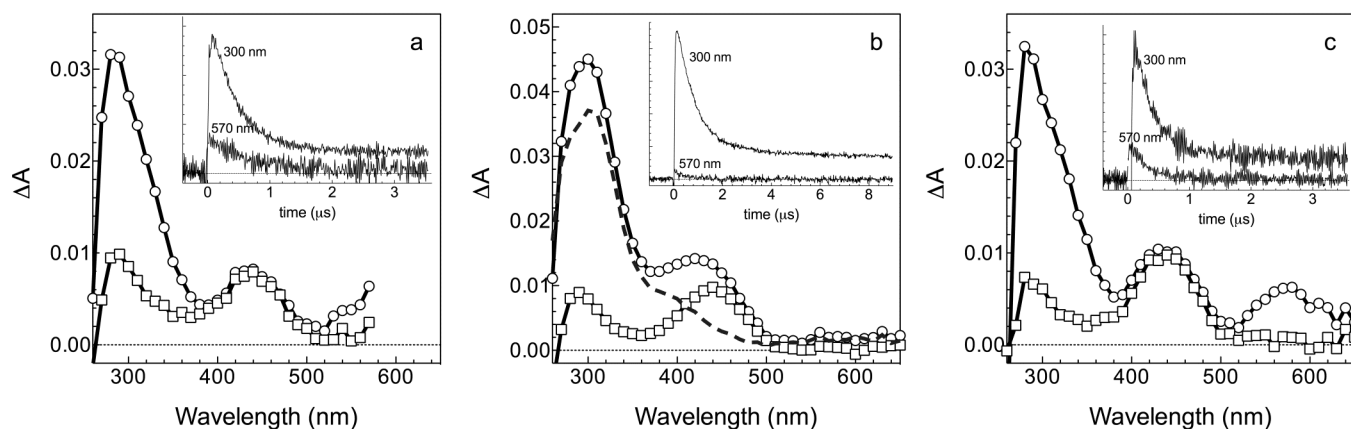


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