

Supporting Information for
Direct Detection, Dimerization and Chemical Trapping of Dimethyl- and
Diphenylstannylene from Photolysis of Stannacyclopent-3-enes in Solution.

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Table of Contents

Figure S1. ^1H NMR spectra of a deaerated 0.04 M solution of 2 in C_6D_{12} (a) before and (b) after 10 minutes photolysis with 254 nm light. The insets in B show an expansion of the δ 0.23-0.48 region of the spectrum and the portion of the $^{119}\text{Sn}[^1\text{H}]$ spectrum containing product peaks.	S5
Figure S2. Concentration vs. time plots for the photolysis of the solution of Fig. S1. The initial slopes of the three plots are 2 , -0.84 ± 0.04 ; 4a , 0.82 ± 0.05 ; $[\text{SnMe}_2]_n$ (δ 0.407) 0.75 ± 0.07 (units, mM min^{-1}).	S5
Figure S3. ^1H NMR spectra of a deaerated 0.04 M solution of 3 in C_6D_{12} (a) before and (b) after 8.3 minutes photolysis with 254 nm light. The resonances marked with ● disappeared after allowing the photolysed solution to stand for 18 hours in the dark.	S6
Figure S4. Concentration vs. time plots for photolysis of ca. 0.04 M solutions of 3 in C_6D_{12} , (a) deaerated (slopes (in units of mM min^{-1}): 3 , -0.32 ± 0.02 ; 4b , 0.30 ± 0.01 ; c- $(\text{SnPh}_2)_6$, 0.0057 ± 0.0005); (b) air-saturated (slopes: 3 , -0.52 ± 0.01 ; 4b , 0.55 ± 0.02).	S7
Figure S5. ^1H NMR spectra of a deaerated 0.04 M solution of 2 in C_6D_{12} containing Me_2SnCl_2 (0.031 M) (a) before and (b) after 10 minutes photolysis with 254 nm light. The inset in B shows the $^{119}\text{Sn}[^1\text{H}]$ NMR spectrum of the photolyzed mixture.	S8
Figure S6. ^1H NMR spectra of an undeaerated 0.04 M solution of 3 in C_6D_{12} containing Me_2SnCl_2 (0.037 M) (a) before, (b) after 2.5 minutes, and (c) after 6.7 minutes photolysis with 254 nm light. No attempt was made to replenish the air in the photolyzate as the experiment proceeded.	S9
Equilibrium Constants for Interconversion of Cyclodistannoxanes 6 , 12 , and 11 (CDCl_3 , 22 °C).	S10

Figure S7. Plots of the concentration ratios of cyclodistannoxanes **6**, **11**, and **12** (i.e. [6]/[12] and [12]/[11]) vs. the dichlorostannane concentration ratio [Me₂SnCl₂]/[8], measured from the ¹H NMR spectra of a ca. 0.012 M solution of **6** in CDCl₃ to which sequential portions of Ph₂SnCl₂ (**8**) and Me₂SnCl₂ were added at ca. 22 °C. The solid lines are the linear least squares fits of the data to [6]/[12] = K_{12_↔6}[Me₂SnCl₂]/[8] (○) and [12]/[11] = K_{11_↔12}[Me₂SnCl₂]/[8] (□); errors are quoted as the standard errors from the least squares analysis. S10

Figure S8. Partial ¹H NMR spectra of a deaerated 0.038 M solution of **3** in C₆D₁₂ containing Me₂SnCl₂ (0.034 M) (a) before, (b) after 2.5 minutes, and (c) after 6.7 minutes photolysis. S11

Figure S9. Concentration vs. time plots for the photolysis of air-saturated C₆D₁₂ solutions of (a) **2** and (b) **3** containing ca. 0.04 M Me₂SnCl₂, and of (c) a deoxygenated C₆D₁₂ solution of **15** containing 0.05 M MeOH; all three solutions also contained Si₂Me₆ (ca. 0.01 M) as internal integration standard. The initial slopes of the plots for the various compounds are (in mM min⁻¹): (a) **2**, -1.06 ± 0.09; Me₂SnCl₂, -1.16 ± 0.05; **4a**, 1.01 ± 0.04; **6**, 0.51 ± 0.01 (A solid, presumed to be **6**, began to precipitate halfway through the experiment, so only the first 4 points were used to evaluate the yield of this product); (b) **3**, -0.97 ± 0.07; Me₂SnCl₂, -1.74 ± 0.40; **4b**, 0.83 ± 0.01; **6**, 0.38 ± 0.02; **8**, 0.76 ± 0.03; **12**, 0.03 ± 0.01; (c) **14**, -0.781 ± 0.002; **4b**, 0.713 ± 0.006; **15**, 0.716 ± 0.012. S12

Figure S10. (a) Transient UV-vis absorption spectra from laser flash photolysis of a rapidly flowed, deoxygenated solution of **3** (7×10^{-4} M) in anhydrous hexanes at 25 °C, recorded over a longer timescale than that shown in Figure 4 of the paper. The spectra were recorded 0.64 – 0.96 μs (○) and 81.1 – 81.9 μs (Δ) after the pulse, using a Pyrex filter in the monitoring beam at wavelengths above 310 nm; the inset shows absorbance vs. time profiles recorded at 340 and 500 nm. (b) Transient absorbance-time profiles recorded for a flowed solution of **3** in deoxygenated hexanes, under similar conditions to those used for the experiment shown in (a) – the sample was different, however. The 340 nm ΔA-time profile was recorded as in (a) and is the average of 10 laser shots, while the 650 nm profile was recorded with a 520 nm cutoff filter (Corning 3-69) in the monitoring beam to filter out overtone absorptions, and is the average of 70 laser shots. The ΔA-time S13

profile at 500 nm (recorded with a Pyrex filter) was quite similar to that obtained in the experiment of (a).

Figure S11. Plots of initial transient absorbance ($(\Delta A)_0$) vs. laser pulse energy from optically matched (at 248 nm), deoxygenated hexanes solutions of (a) benzophenone and **2**, and (b) benzophenone and **3**, for determination of the extinction coefficients of the SnMe_2 and SnPh_2 absorption bands at 500 nm. The benzophenone triplet (${}^3\text{BP}$; $\Phi = 1.0$) was monitored at 525 nm ($\epsilon = 6,250 \pm 1,250 \text{ M}^{-1}\text{cm}^{-1}$).¹ The slopes of the plots are (a) ${}^3\text{BP}$, $(6.0 \pm 0.1) \times 10^{-4}$, SnMe_2 $(1.43 \pm 0.02) \times 10^{-4}$; (b) ${}^3\text{BP}$, $(4.61 \pm 0.08) \times 10^{-4}$, SnPh_2 $(1.13 \pm 0.03) \times 10^{-4}$.

Figure S12. (a) Time-resolved UV-vis spectra recorded by laser photolysis of SnMe_2 precursor **2** in hexanes containing 7 mM MeOH, 0.22-0.29 μs (\circ), 1.25-1.31 μs (\square) and 17.5-17.7 μs (Δ) after the laser pulse (25 °C), and absorbance-time profiles at selected wavelengths (inset). (b) Plot of $(\Delta A_0)_0 / (\Delta A_0)_Q$ for complexation of SnMe_2 with MeOH in hexanes at 25 °C; the solid line is the linear least squares fit of the data to equation 14.

Computational Studies

Table S1. Calculated Electronic Energies, Standard Enthalpies (298.15 K), and Standard Free Energies (298.15 K) of Stationary Points in the Dimerization of SnMe_2 and SnPh_2 , Calculated at the $\omega\text{B97X}/6-31+\text{G(d,p)}^{\text{C,H,O}}\text{-LANL2DZdp}^{\text{Sn}}$ Level of Theory Relative to the Isolated Reactants (in kcal mol⁻¹).

Figure S13. Selected geometric parameters, electronic energies and standard free energies (in parentheses) for SnPh_2 and the SnPh_2 -dimers **16b**, **17b**, **20** and (*trans*)-**21**, calculated at the $\omega\text{B97XD}/6-31+\text{G(d,p)}^{\text{C,H,O}}\text{-LANL2DZdp}^{\text{Sn}}$ level (hydrogens omitted for clarity).

Figure S14. Plots of ΔE vs. geometry from relaxed PES scans of (A) the C1-Sn2-Sn3-C4 dihedral angle in **16b** and (B) the Sn-C bond distance involving the bridging phenyl group in stannylenestannylene **20**, carried out at the $\omega\text{B97XD}/6-31+\text{G(d,p)}^{\text{C,H,O}}\text{-LANL2DZdp}^{\text{Sn}}$ level of theory.

Figure S15. Plot of calculated relative electronic energies (ΔE ; relative to two SnPh_2 moieties at infinite separation) vs. Sn-Sn bond distance, from relaxed potential energy

surface scans of the Sn-Sn bond distances ($d_{\text{Sn-Sn}}$) in **17b** and **20** at the $\omega\text{B97XD}/6-31+\text{G}(\text{d},\text{p})^{\text{C},\text{H},\text{O}}\text{-LANL2DZdp}^{\text{Sn}}$ level of theory. The calculated structures at various $d_{\text{Sn-Sn}}$ values in the calculations are also shown.

Calculated Structures and Energies (in Hartrees)	S19
Dimethylstannylene	S19
Tetramethyldistannene (16a)	S20
Methyltrimethylstannylstannylene (17a)	S21
Diphenylstannylene	S22
Tetraphenyldistannene (16b)	S23
Phenyltriphenylstannylstannylene (17b)	S24
Singly-bridged SnPh_2 -dimer (20)	S25
Doubly-bridged SnPh_2 -dimer (21)	S26
Transition State 28^\ddagger (17b \leftrightarrow 25[‡] \leftrightarrow 21)	S28
Transition State 29^\ddagger (20 \leftrightarrow 26[‡] \leftrightarrow 17b)	S29
Methanol	S30
$\text{SnMe}_2\text{-MeOH}$ Complex (18)	S31
$\text{SnPh}_2\text{-MeOH}$ Complex (19)	S32
Figure S16. (a) ^1H and (b) DEPTq NMR spectra of 2 in CDCl_3 . The inset in (a) contains an expansion of the spectrum to show contributions from impurities.	S33
References	S34

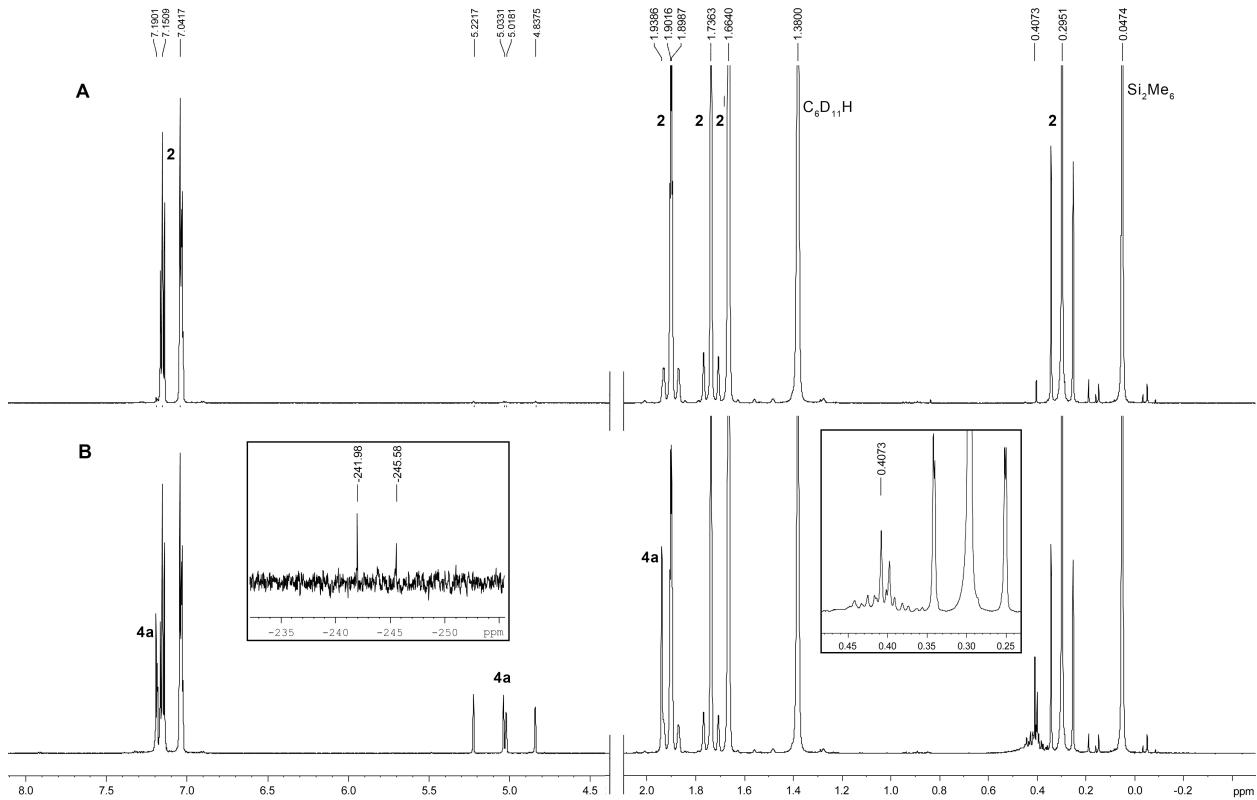


Figure S1. ¹H NMR spectra of a deaerated 0.04 M solution of **2** in C₆D₁₂ (a) before and (b) after 10 minutes photolysis with 254 nm light. The insets in B show an expansion of the δ 0.23–0.48 region of the spectrum and the portion of the ¹¹⁹Sn[¹H] spectrum containing product peaks.

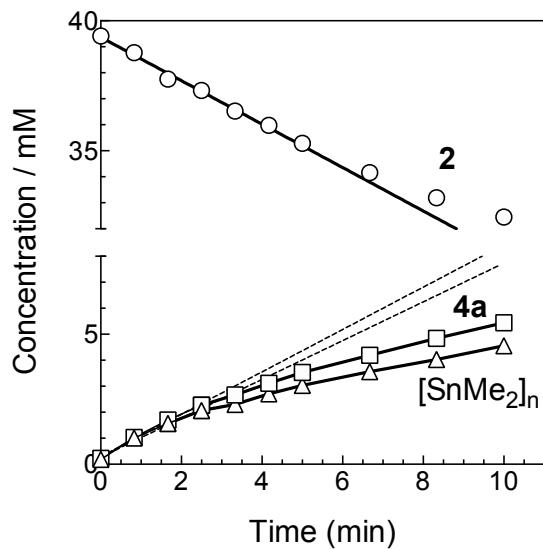


Figure S2. Concentration vs. time plots for the photolysis of the solution of Fig. S1. The initial slopes of the three plots are **2**, -0.84 ± 0.04 ; **4a**, 0.82 ± 0.05 ; $[\text{SnMe}_2]_n$ ($\delta 0.407$) 0.75 ± 0.07 (units, mM min⁻¹).

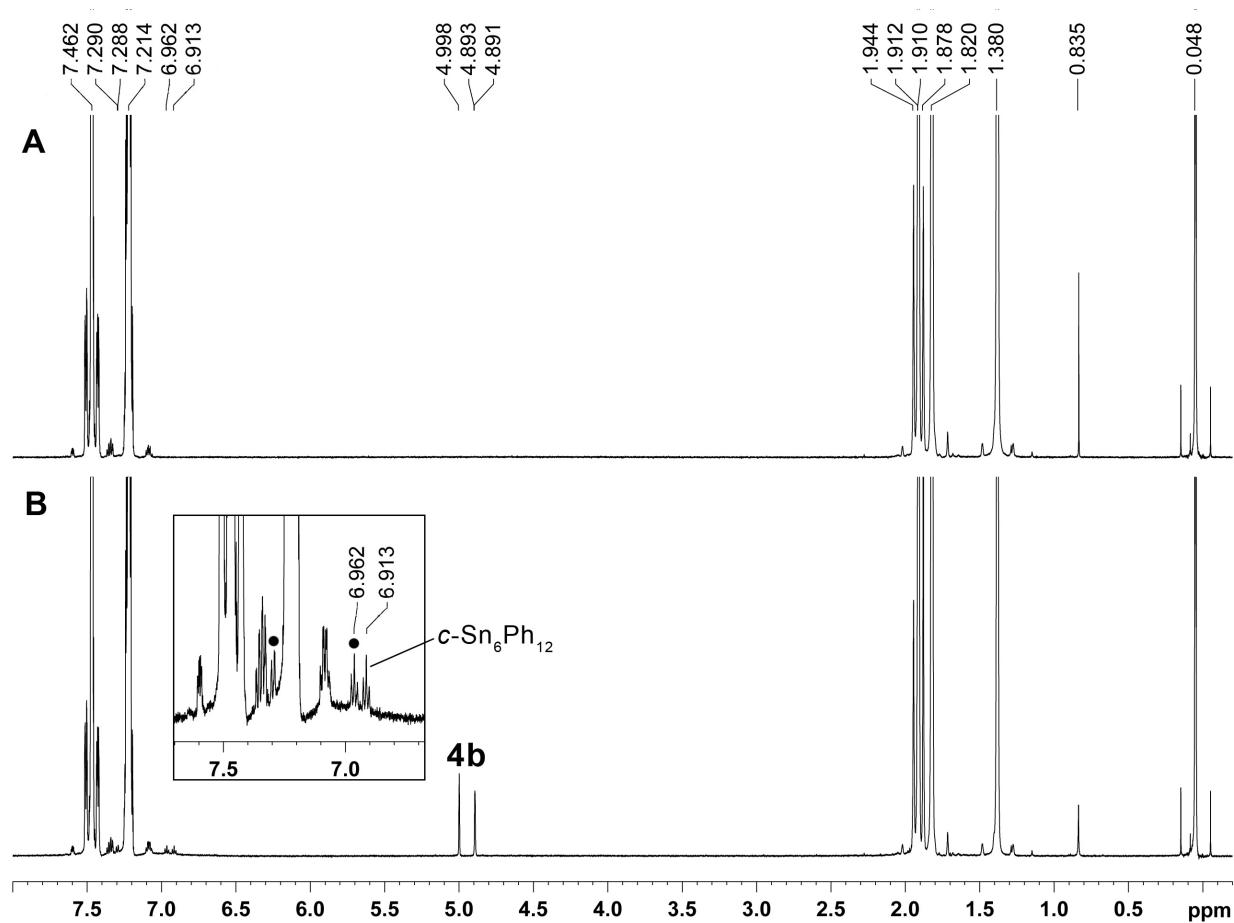
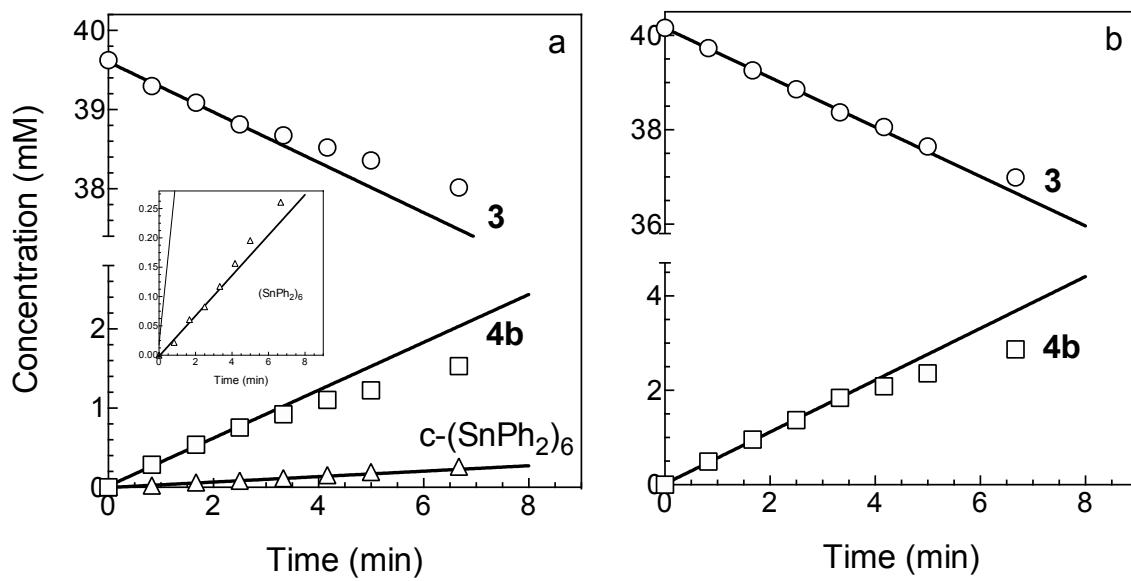


Figure S3. ¹H NMR spectra of a deaerated 0.04 M solution of **3** in C₆D₁₂ (a) before and (b) after 8.3 minutes photolysis with 254 nm light. The resonances marked with ● disappeared after allowing the photolysed solution to stand for 18 hours in the dark.



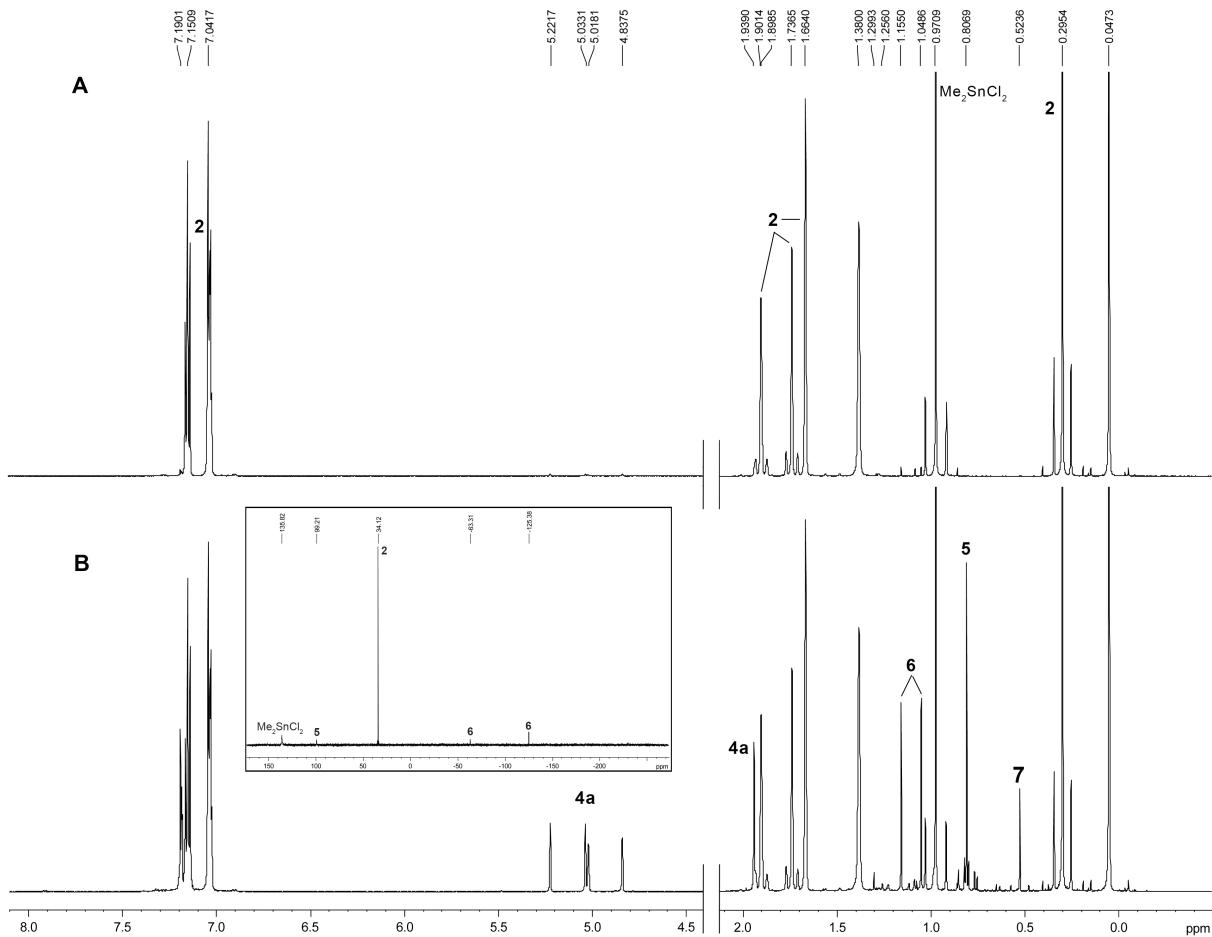


Figure S5. ¹H NMR spectra of a deaerated 0.04 M solution of **2** in C₆D₁₂ containing Me₂SnCl₂ (0.031 M) (a) before and (b) after 10 minutes photolysis with 254 nm light. The inset in B shows the ¹¹⁹Sn[¹H] NMR spectrum of the photolyzed mixture.

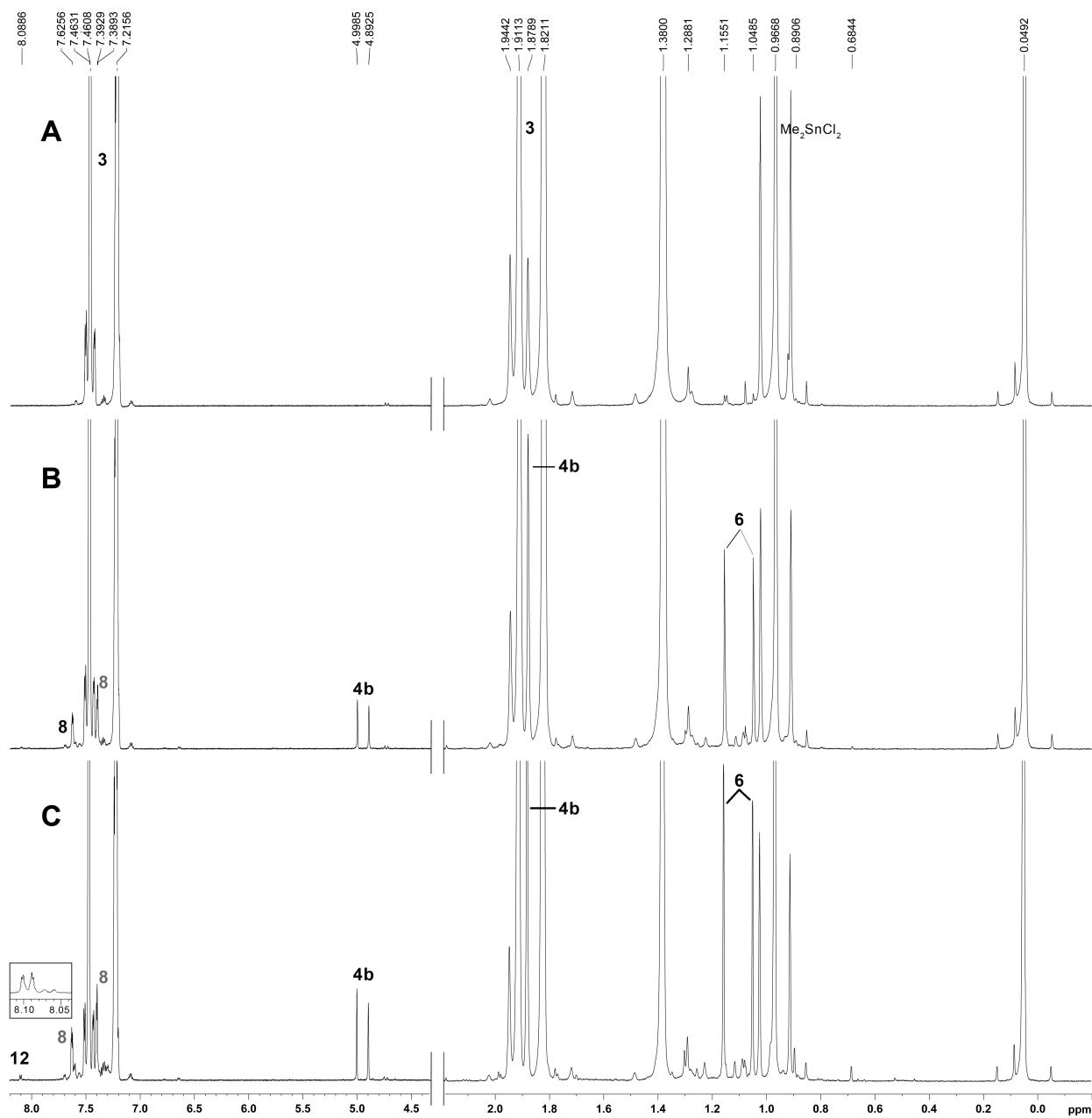


Figure S6. ¹H NMR spectra of an undeaerated 0.04 M solution of **3** in C_6D_{12} containing Me_2SnCl_2 (0.037 M) (a) before, (b) after 2.5 minutes, and (c) after 6.7 minutes photolysis with 254 nm light. No attempt was made to replenish the air in the photolyzate as the experiment proceeded.

Equilibrium Constants for Interconversion of Cyclodistannoxanes **6**, **12**, and **11** (CDCl_3 , 22 °C).

A mixture of **6** (13.7 mg, 11.9 mM) and hexamethyldisilane (4.00 μL , 13.0 mM) in CDCl_3 (1.50 mL) was ultrasonicated in a 5 mL screw cap glass vial at 22 °C, and then the $^1\text{H-NMR}$ spectrum of the resulting opaque solution was recorded. The NMR sample was recombined with the bulk of the solution, **8** (24.2 mg, 46.9 mM) was added, the solution was shaken briefly, and the $^1\text{H-NMR}$ spectrum of the resulting clear colourless solution (solution ‘a’) was recorded. The procedure was repeated with an additional amount of **8** (26.7 mg, 51.8 mM; solution ‘b’), and again with two successive additions of Me_2SnCl_2 (31.8 mg, 96.5 mM (solution ‘c’) and 31.5 mg, 95.6 mM (solution ‘d’)).

The relative concentrations of the five components of each of the mixtures (see Table below) were calculated from the NMR integrals, using the signals at δ 1.19 (s, 12H), δ 8.06 (d, 4H), δ 8.03 (d, 8H), δ 1.21 (s, 6H), and δ 7.71 (d, 4H) for **6**, **12**, **11**, Me_2SnCl_2 , and Ph_2SnCl_2 (**8**), respectively. The equilibrium constants $K_{11 \leftrightarrow 12}$ and $K_{12 \leftrightarrow 6}$ were then calculated by least squares analysis of plots of the concentration ratios of the cyclodistannoxanes vs. those of the dichlorostannanes (Fig. S7).

Quantities Mixed / mM						
Solution	6	Me_2SnCl_2	8	$([\mathbf{6}]/[\mathbf{12}])_e$	$([\mathbf{12}]/[\mathbf{11}])_e$	$([\text{Me}_2\text{SnCl}_2]/[\mathbf{8}])_e$
(a)	11.9	0	46.9	0.266	0.758	0.383
(b)	11.9	0	98.7	0.205	0.560	0.191
(c)	11.9	96.5	98.7	0.780	1.90	1.32
(d)	11.9	192	98.7	1.23	2.43	2.45

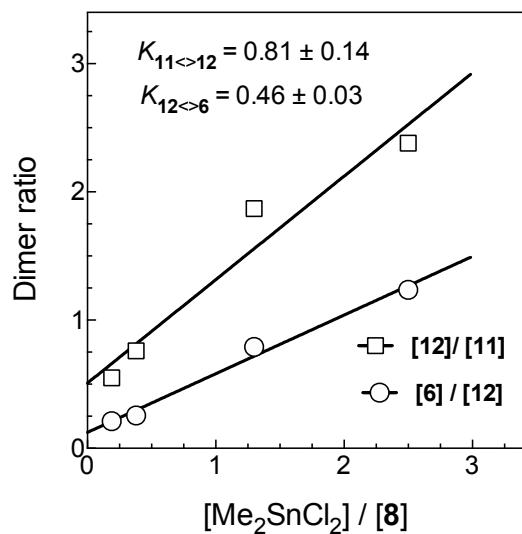


Figure S7. Plots of the concentration ratios of cyclodistannoxanes **6**, **11**, and **12** (i.e. $[\mathbf{6}]/[\mathbf{12}]$ and $[\mathbf{12}]/[\mathbf{11}]$) vs. the dichlorostannane concentration ratio $[\text{Me}_2\text{SnCl}_2]/[\mathbf{8}]$, measured from the ^1H NMR spectra of a ca. 0.012 M solution of **6** in CDCl_3 to which sequential portions of Ph_2SnCl_2 (**8**) and Me_2SnCl_2 were added at ca. 22 °C. The solid lines are the linear least squares fits of the data to $[\mathbf{6}]/[\mathbf{12}] = K_{12 \leftrightarrow 6}[\text{Me}_2\text{SnCl}_2]/[\mathbf{8}]$ (○) and $[\mathbf{12}]/[\mathbf{11}] = K_{11 \leftrightarrow 12}[\text{Me}_2\text{SnCl}_2]/[\mathbf{8}]$ (□); errors are quoted as the standard errors from the least squares analysis.

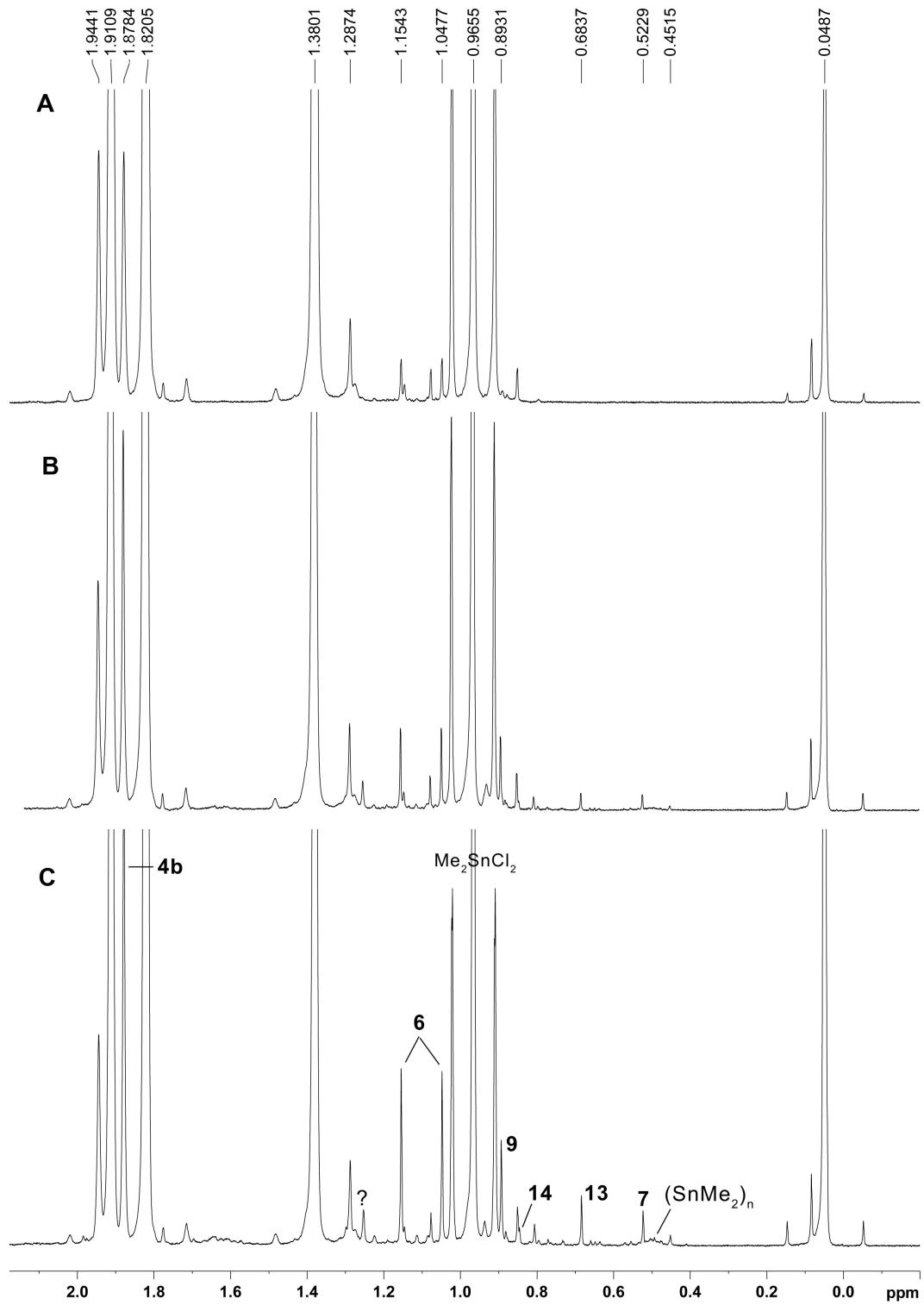


Figure S8. Partial ¹H NMR spectra of a deaerated 0.038 M solution of **3** in C₆D₁₂ containing Me₂SnCl₂ (0.034 M) (a) before, (b) after 2.5 minutes, and (c) after 6.7 minutes photolysis.

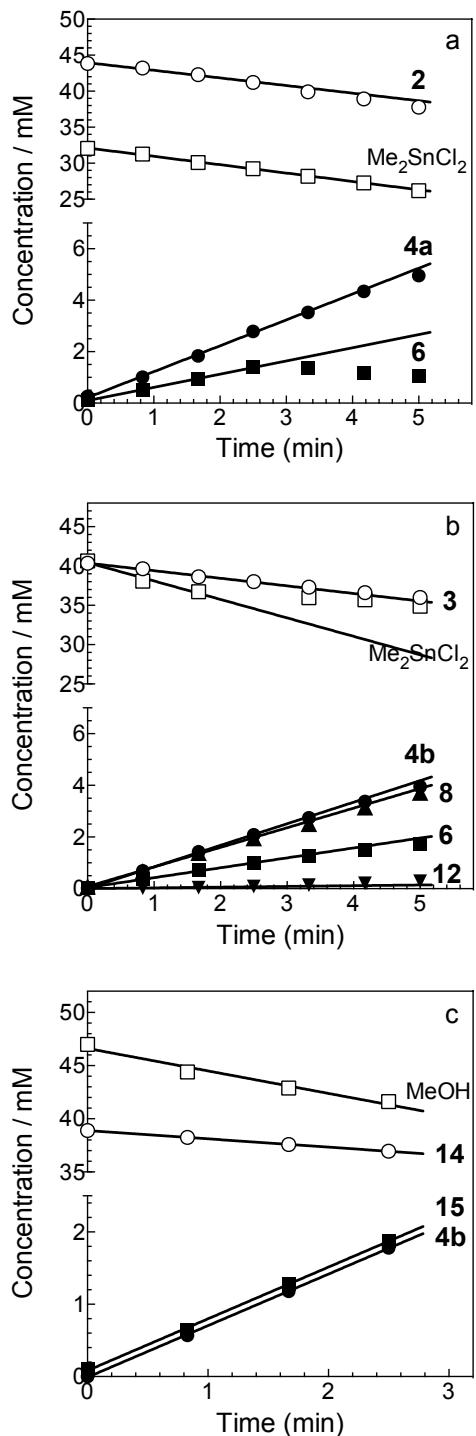


Figure S9. Concentration vs. time plots for the photolysis of air-saturated C_6D_{12} solutions of (a) **2** and (b) **3** containing ca. 0.04 M Me_2SnCl_2 , and of (c) a deoxygenated C_6D_{12} solution of **15** containing 0.05 M MeOH; all three solutions also contained Si_2Me_6 (ca. 0.01 M) as internal integration standard. The initial slopes of the plots for the various compounds are (in mM min^{-1}): (a) **2**, -1.06 ± 0.09 ; Me_2SnCl_2 , -1.16 ± 0.05 ; **4a**, 1.01 ± 0.04 ; **6**, 0.51 ± 0.01 (A solid, presumed to be **6**, began to precipitate halfway through the experiment, so only the first 4 points were used to evaluate the yield of this product); (b) **3**, -0.97 ± 0.07 ; Me_2SnCl_2 , -1.74 ± 0.40 ; **4b**, 0.83 ± 0.01 ; **6**, 0.38 ± 0.02 ; **8**, 0.76 ± 0.03 ; **12**, 0.03 ± 0.01 ; (c) **14**, -0.781 ± 0.002 ; **4b**, 0.713 ± 0.006 ; **15**, 0.716 ± 0.012 .

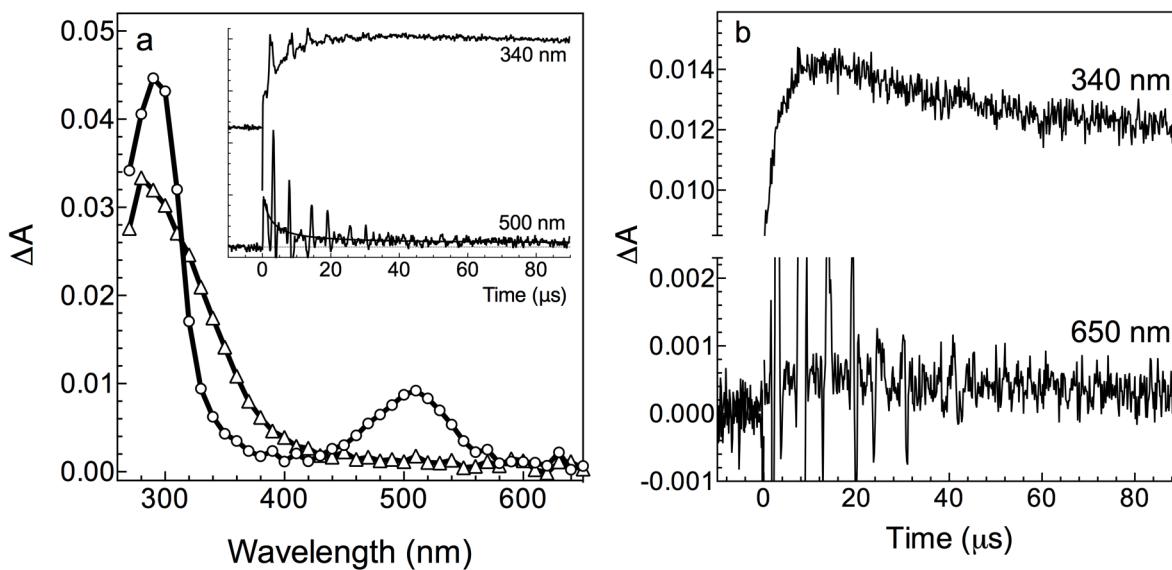


Figure S10. (a) Transient UV-vis absorption spectra from laser flash photolysis of a rapidly flowed, deoxygenated solution of **3** (7×10^{-4} M) in anhydrous hexanes at 25 °C, recorded over a longer timescale than that shown in Figure 5 of the paper. The spectra were recorded 0.64 – 0.96 μ s (○) and 81.1 – 81.9 μ s (Δ) after the pulse, using a Pyrex filter in the monitoring beam at wavelengths above 310 nm; the inset shows absorbance vs. time profiles recorded at 340 and 500 nm. (b) Transient absorbance-time profiles recorded for a flowed solution of **3** in deoxygenated hexanes, under similar conditions to those used for the experiment shown in (a) – the sample was different, however. The 340 nm ΔA -time profile was recorded as in (a) and is the average of 10 laser shots, while the 650 nm profile was recorded with a 520 nm cutoff filter (Corning 3-69) in the monitoring beam to filter out overtone absorptions, and is the average of 70 laser shots. The ΔA -time profile at 500 nm (recorded with a Pyrex filter) was quite similar to that obtained in the experiment of (a).

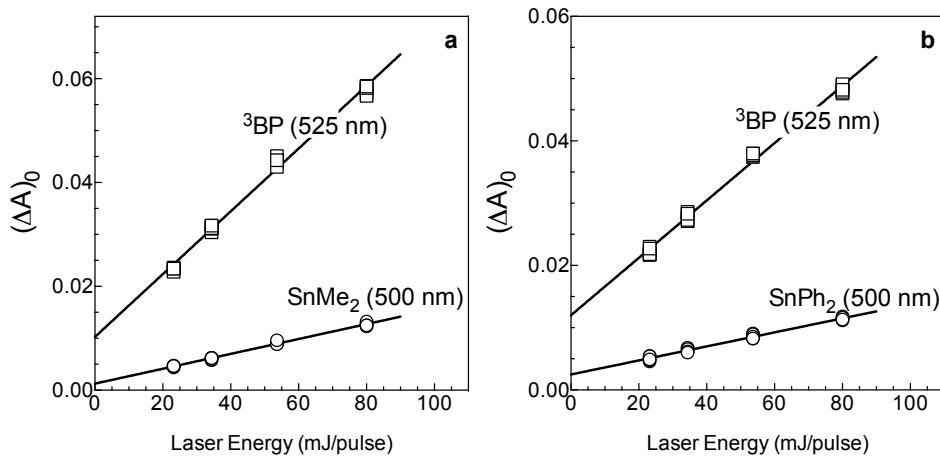


Figure S11. Plots of initial transient absorbance $((\Delta A)_0)$ vs. laser pulse energy from optically matched (at 248 nm), deoxygenated hexanes solutions of (a) benzophenone and **2**, and (b) benzophenone and **3**, for determination of the extinction coefficients of the SnMe_2 and SnPh_2 absorption bands at 500 nm. The benzophenone triplet (${}^3\text{BP}$; $\Phi = 1.0$) was monitored at 525 nm ($\epsilon = 6,250 \pm 1,250 \text{ M}^{-1}\text{cm}^{-1}$).¹ The slopes of the plots are (a) ${}^3\text{BP}$, $(6.0 \pm 0.1) \times 10^{-4}$, SnMe_2 $(1.43 \pm 0.02) \times 10^{-4}$; (b) ${}^3\text{BP}$, $(4.61 \pm 0.08) \times 10^{-4}$, SnPh_2 $(1.13 \pm 0.03) \times 10^{-4}$.

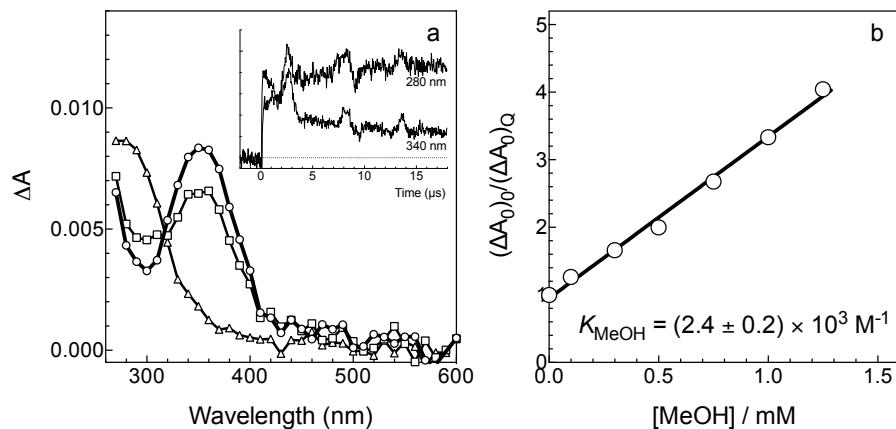


Figure S12. (a) Time-resolved UV-vis spectra recorded by laser photolysis of SnMe_2 precursor **2** in hexanes containing 7 mM MeOH, 0.22-0.29 μs (\circ), 1.25-1.31 μs (\square) and 17.5-17.7 μs (Δ) after the laser pulse (25 °C), and absorbance-time profiles at selected wavelengths (inset). (b) Plot of $(\Delta A_0)_0 / (\Delta A_0)_Q$ for complexation of SnMe_2 with MeOH in hexanes at 25 °C; the solid line is the linear least squares fit of the data to equation 14.

Computational Studies

Theoretical calculations were carried out using the Gaussian09 (Rev. B.01) suite of programs.² Energies in the tables of calculated structures and energies are reported in units of Hartree/Particle.

Table S1. Calculated Electronic Energies, Standard Enthalpies (298.15 K), and Standard Free Energies (298.15 K) of Stationary Points in the Dimerization of SnMe₂ and SnPh₂, Calculated at the ωB97X/6-31+G(d,p)^{C,H,O}-LANL2DZdp^{Sn} Level of Theory Relative to the Isolated Reactants (in kcal mol⁻¹).

Species	ΔE_{elec}	ΔH°	ΔG°	$\Delta(\Delta H^\circ)^a$	$\Delta(\Delta G^\circ)^a$
Me ₂ Sn=SnMe ₂ (16a)	-23.4	-20.7	-10.4	-1.4	-1.5
MeSnSnMe ₃ (17a)	-32.2	-29.8	-20.3	-0.5	-0.8
Ph ₂ Sn=SnPh ₂ (16b)	-19.5	-18.0	-8.3	-3.1	-3.1
PhSnSnPh ₃ (17b)	-30.8	-29.3	-20.4	-3.7	-0.7
PhSn(C ₆ H ₅)SnPh ₂ (23)	-24.2	-22.9	-11.1	-5.4	-5.2
<i>trans</i> -PhSn(C ₆ H ₅) ₂ SnPh (24)	-14.0	-12.6	-0.2	-5.4	-5.6
Transition state 28 [‡]	+8.1	+8.6	+21.7	-7.6	-7.3
Transition state 29 [‡]	-24.2	-23.5	-10.3	-5.0	-5.4
Me ₂ Sn-O(H)Me (19a) ^b	-14.0	-12.5	-2.6	+0.7	+0.7
Ph ₂ Sn-O(H)Me (19b) ^b	-15.9	-14.4	-4.0	+0.1	+0.7

a. Defined as $\Delta(\Delta E) = \Delta E(\omega B97XD) - \Delta E(\omega B97X)$.

b. Corrected for Basis Set Superposition Error (BSSE).

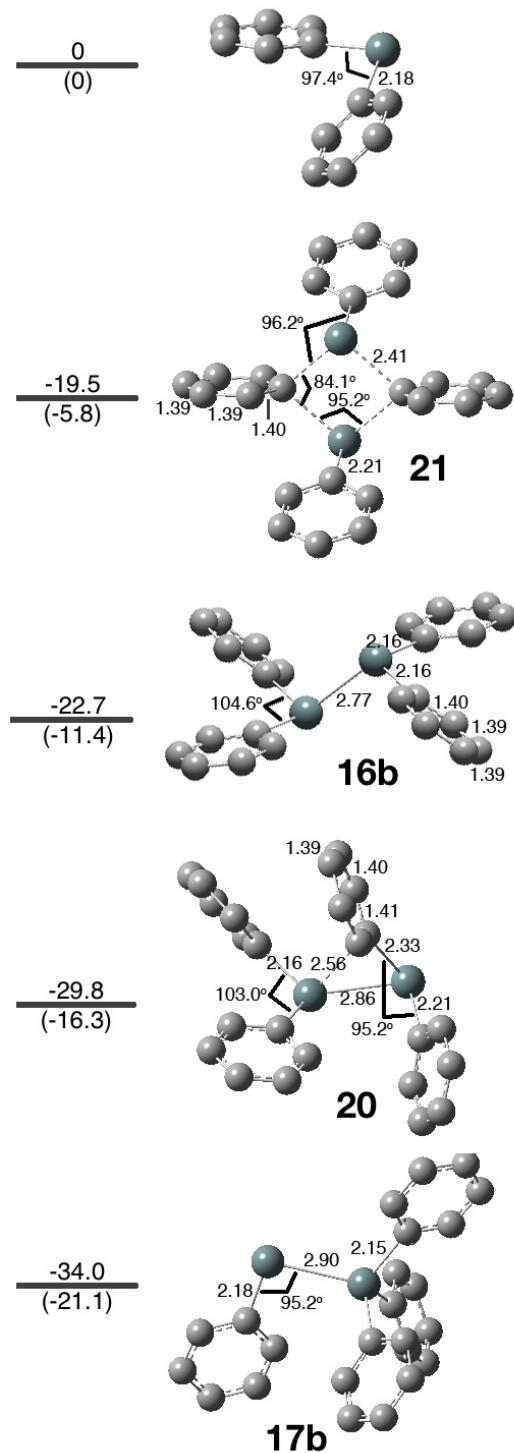


Figure S13. Selected geometric parameters, electronic energies and standard free energies (in parentheses) for SnPh_2 and the SnPh_2 -dimers **16b**, **17b**, **20** and (*trans*)-**21**, calculated at the ω B97XD/6-31+G(d,p)^{C,H,O}-LANL2DZdp^{Sn} level (hydrogens omitted for clarity).

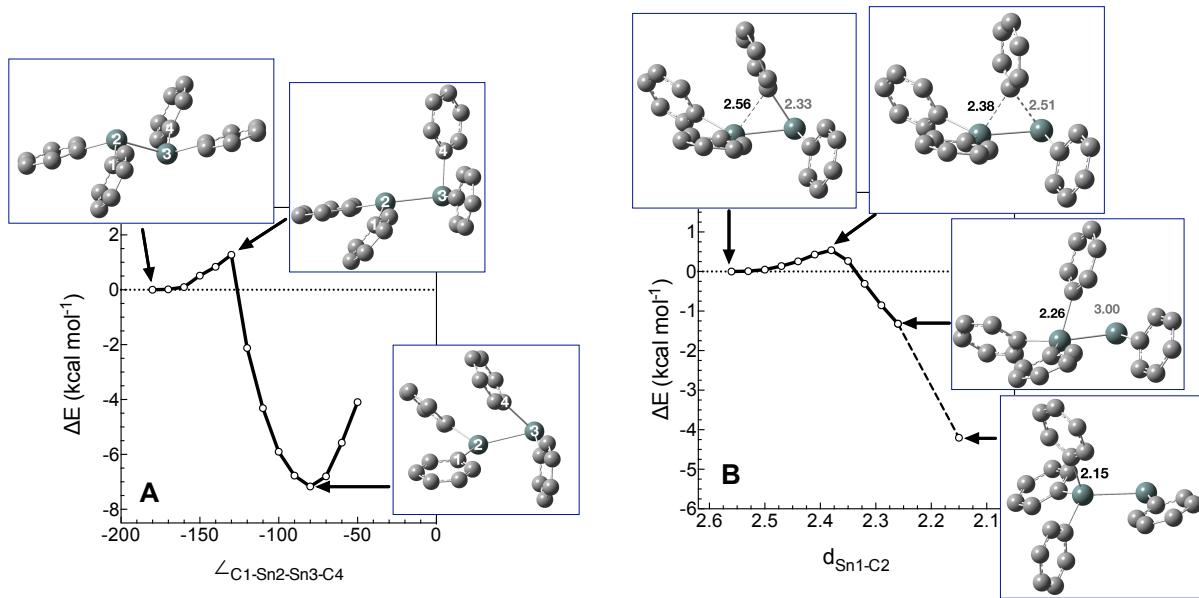


Figure S14. Plots of ΔE vs. geometry from relaxed PES scans of (A) the C1-Sn2-Sn3-C4 dihedral angle in **16b** and (B) the Sn-C bond distance involving the bridging phenyl group in stannylenestannylene **20**, carried out at the ω B97XD/6-31+G(d,p)^{C,H,O}-LANL2DZdp^{Sn} level of theory.

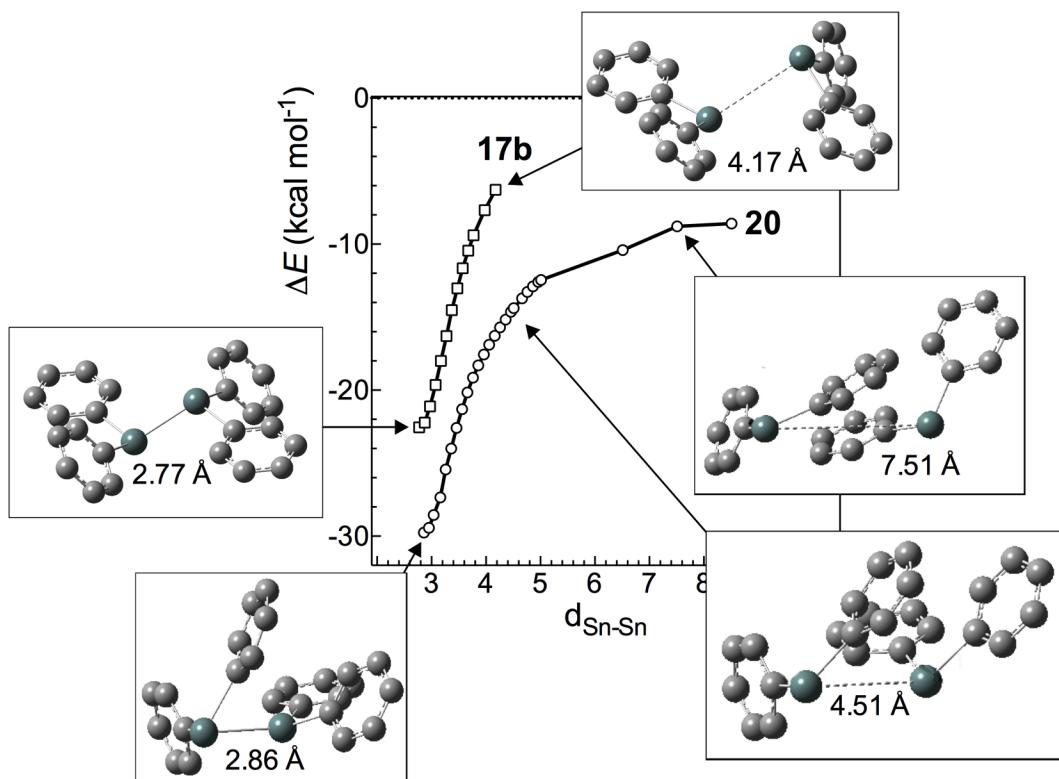
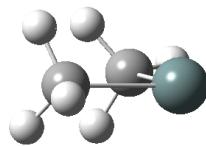


Figure S15. Plot of calculated relative electronic energies (ΔE ; relative to two SnPh_2 moieties at infinite separation) vs. Sn-Sn bond distance, from relaxed potential energy surface scans of the Sn-Sn bond distances ($d_{\text{Sn-Sn}}$) in **17b** and **20** at the $\omega\text{B97XD}/6-31+\text{G}(\text{d},\text{p})^{\text{C},\text{H},\text{O}}\text{-LANL2DZdp}^{\text{Sn}}$ level of theory. The calculated structures at various $d_{\text{Sn-Sn}}$ values in the calculations are also shown.

Calculated structures and energies (in Hartrees)

Dimethylstannylene [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

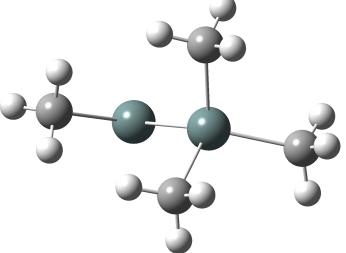
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Thermal correction to Energy=	0.076414
Thermal correction to Enthalpy=	0.077358
Thermal correction to Gibbs Free Energy=	0.041357
Sum of electronic and zero-point Energies=	-83.090800
Sum of electronic and thermal Energies=	-83.084939
Sum of electronic and thermal Enthalpies=	-83.083995
Sum of electronic and thermal Free Energies=	-83.119996
Sn	0.00000000 0.00000000 0.42625400
C	0.00000000 1.58572500 -1.08795300
C	0.00000000 -1.58572500 -1.08795300
H	0.78719500 -1.40335700 -1.83008100
H	-0.95629800 -1.55642000 -1.62796800
H	-0.12840600 2.58802300 -0.67058000
H	-0.78719500 1.40335700 -1.83008100
H	0.95629800 1.55642000 -1.62796800
H	0.12840600 -2.58802300 -0.67058000



Tetramethyldistannene (16a) [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

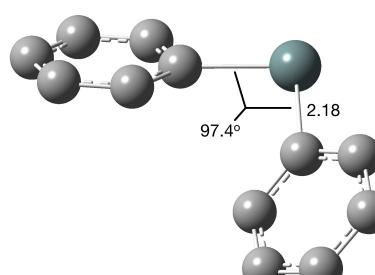
Zero-point correction=	0.145100			
Thermal correction to Energy=	0.158137			
Thermal correction to Enthalpy=	0.159081			
Thermal correction to Gibbs Free Energy=	0.103332			
Sum of electronic and zero-point Energies=	-166.217246			
Sum of electronic and thermal Energies=	-166.204209			
Sum of electronic and thermal Enthalpies=	-166.203265			
Sum of electronic and thermal Free Energies=	-166.259014			
Sn	-0.26578200	-0.00348500	-1.34599500	
C	0.61640400	-1.65746700	-2.44423400	
C	0.48257600	1.72475300	-2.42881600	
H	1.55748100	1.62122000	-2.60346700	
H	-0.01648300	1.78750700	-3.40132000	
H	0.49613700	-2.60419100	-1.91208300	
H	0.12884200	-1.74541500	-3.42064300	
H	1.68185100	-1.47361900	-2.61091200	
H	0.29645300	2.65381900	-1.88463400	
Sn	0.26578200	0.00348500	1.34599500	
C	-0.61640400	1.65746700	2.44423400	
C	-0.48257600	-1.72475300	2.42881600	
H	-1.55748100	-1.62122000	2.60346700	
H	0.01648300	-1.78750700	3.40132000	
H	-0.49613700	2.60419100	1.91208300	
H	-0.12884200	1.74541500	3.42064300	
H	-1.68185100	1.47361900	2.61091200	
H	-0.29645300	-2.65381900	1.88463400	

Methyltrimethylstannylstannylene (17a) [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

Zero-point correction=	0.144310	
Thermal correction to Energy=	0.157597	
Thermal correction to Enthalpy=	0.158541	
Thermal correction to Gibbs Free Energy=	0.101082	
Sum of electronic and zero-point Energies=	-166.230467	
Sum of electronic and thermal Energies=	-166.217180	
Sum of electronic and thermal Enthalpies=	-166.216236	
Sum of electronic and thermal Free Energies=	-166.273695	
Sn	1.77332200	
Sn	-1.10036500	
C	2.09377800	
H	1.63502400	
H	1.60925000	
H	3.15607200	
C	-1.67517600	
H	-1.47950300	
H	-2.74587900	
H	-1.12783400	
C	-2.32072400	
H	-3.37989800	
H	-2.17176100	
H	-2.07386500	
C	-1.70144900	
H	-2.77971800	
H	-1.47985000	
H	-1.18845400	
	1.93847700	
	1.94476400	

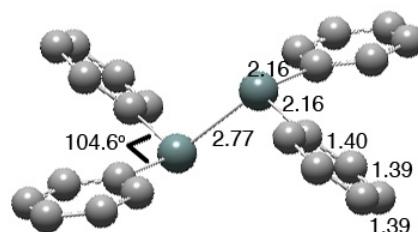
Diphenylstannylene [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

Zero-point correction=	0.181717		
Thermal correction to Energy=	0.193336		
Thermal correction to Enthalpy=	0.194280		
Thermal correction to Gibbs Free Energy=	0.141738		
Sum of electronic and zero-point Energies=	-466.330374		
Sum of electronic and thermal Energies=	-466.318756		
Sum of electronic and thermal Enthalpies=	-466.317811		
Sum of electronic and thermal Free Energies=	-466.370353		
C	0.00000000	1.63830700	0.01432400
C	0.00000000	-1.63830700	0.01432400
C	-0.60552700	1.56949800	-1.25305800
C	-0.64048200	2.67516700	-2.10122900
C	-0.05641700	3.87766800	-1.70441600
C	0.54637600	3.97520800	-0.45181500
C	0.55786200	2.87012400	0.39899700
C	0.60552700	-1.56949800	-1.25305800
C	-0.55786200	-2.87012400	0.39899700
C	0.64048200	-2.67516700	-2.10122900
C	0.05641700	-3.87766800	-1.70441600
C	-0.54637600	-3.97520800	-0.45181500
H	-1.11919600	2.59770600	-3.07330500
H	0.99735100	4.91133200	-0.13537900
H	1.01458200	2.97862600	1.38221100
H	-1.01458200	-2.97862600	1.38221100
H	1.11919600	-2.59770600	-3.07330500
H	-1.05802800	0.63952800	-1.58933300
H	-0.07590400	4.73708800	-2.36829900
H	1.05802800	-0.63952800	-1.58933300
H	0.07590400	-4.73708800	-2.36829900
H	-0.99735100	-4.91133200	-0.13537900
Sn	0.00000000	0.00000000	1.45469200



Tetraphenyldistannene (16b) [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

Zero-point correction=	0.365049
Thermal correction to Energy=	0.390241
Thermal correction to Enthalpy=	0.391186
Thermal correction to Gibbs Free Energy=	0.301573
Sum of electronic and zero-point Energies=	-932.695367
Sum of electronic and thermal Energies=	-932.670174
Sum of electronic and thermal Enthalpies=	-932.669229
Sum of electronic and thermal Free Energies=	-932.758842
Sn	0.01178900 -0.76585000 -1.15671300
C	-1.71492600 -0.38127600 -2.38952100
C	1.70107600 -0.27954500 -2.41638000
C	-1.66179000 -0.42424200 -3.79020100
C	-2.80935600 -0.22956400 -4.55951500
C	-4.03306000 0.01062300 -3.93871900
C	-4.10800500 0.04603100 -2.54643000
C	-2.96075900 -0.15556900 -1.78207600
C	1.64471600 0.73216600 -3.38742200
C	2.92198200 -0.94460400 -2.22799900
C	2.76680700 1.06455800 -4.14497200
C	3.96956400 0.38803900 -3.94622700
C	4.04679700 -0.61847400 -2.98589800
H	-2.74587400 -0.26651900 -5.64329500
H	-5.06072500 0.22574400 -2.05679400
H	-3.04661500 -0.13368500 -0.69661900
H	3.00607700 -1.72923700 -1.47776000
H	2.70355000 1.85327900 -4.88905700
H	-0.71728600 -0.61059100 -4.29614200
H	-4.92703200 0.16500000 -4.53557600
H	0.71801400 1.27576000 -3.55458000
H	4.84410900 0.64635200 -4.53599600
H	4.98211100 -1.14693200 -2.82533100
Sn	-0.01178900 0.76585000 1.15671300
C	1.71492600 0.38127600 2.38952100
C	-1.70107600 0.27954500 2.41638000
C	1.66179000 0.42424200 3.79020100
C	2.80935600 0.22956400 4.55951500
C	4.03306000 -0.01062300 3.93871900
C	4.10800500 -0.04603100 2.54643000
C	2.96075900 0.15556900 1.78207600
C	-1.64471600 -0.73216600 3.38742200
C	-2.92198200 0.94460400 2.22799900
C	-2.76680700 -1.06455800 4.14497200
C	-3.96956400 -0.38803900 3.94622700
C	-4.04679700 0.61847400 2.98589800
H	2.74587400 0.26651900 5.64329500
H	5.06072500 -0.22574400 2.05679400
H	3.04661500 0.13368500 0.69661900
H	-3.00607700 1.72923700 1.47776000

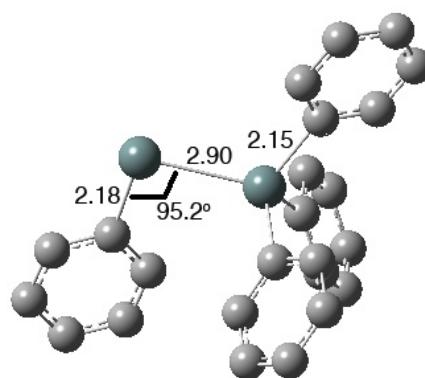


H	-2.70355000	-1.85327900	4.88905700
H	0.71728600	0.61059100	4.29614200
H	4.92703200	-0.16500000	4.53557600
H	-0.71801400	-1.27576000	3.55458000
H	-4.84410900	-0.64635200	4.53599600
H	-4.98211100	1.14693200	2.82533100

Phenyl(triphenylstannylyl)stannylene (17b) [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

Zero-point correction=	0.364756
Thermal correction to Energy=	0.389091
Thermal correction to Enthalpy=	0.390035
Thermal correction to Gibbs Free Energy=	0.304012
Sum of electronic and zero-point Energies=	-932.713531
Sum of electronic and thermal Energies=	-932.689196
Sum of electronic and thermal Enthalpies=	-932.688252
Sum of electronic and thermal Free Energies=	-932.774275

Sn	1.20589500	-1.71538200	-1.44602600
Sn	-0.71701600	-0.02183600	-0.08323900
C	2.95927600	-0.62050900	-0.75190200
C	4.18624600	-1.30755300	-0.82208100
C	5.37386600	-0.73116900	-0.37234400
H	6.30523800	-1.28687000	-0.43058500
C	5.36324600	0.56261900	0.14426500
H	6.28556200	1.01707300	0.49391900
C	4.16583300	1.27512100	0.20593000
H	4.15120200	2.28688800	0.60074100
C	2.98148700	0.68834500	-0.23328700
C	-2.76157500	-0.40992800	-0.61146900
C	-3.20594700	-1.71356000	-0.87508900
C	-4.54202400	-1.97464500	-1.17694500
H	-4.86459400	-2.99291100	-1.37435400
C	-5.46094800	-0.92779500	-1.22575200
H	-6.50266700	-1.12681300	-1.45940800
C	-5.03765200	0.37581000	-0.97313100
H	-5.74994200	1.19510200	-1.01025000
C	-3.69995800	0.63001600	-0.66936600
C	-0.45228700	2.10797900	-0.24916200
C	-0.60973500	2.76305400	-1.47880900
H	-0.89883000	2.20164500	-2.36579300
C	-0.41027600	4.13815100	-1.59521300
H	-0.54047200	4.62570900	-2.55708300
C	-0.04196100	4.88478500	-0.47699700
H	0.11780200	5.95521800	-0.56584900
C	0.11816200	4.25174500	0.75402400
H	0.40010300	4.82865200	1.63039100
C	-0.08653200	2.87599600	0.86500600
H	0.04018200	2.40009300	1.83531500
C	-0.53026300	-0.47634400	2.01339000



C	-1.65554900	-0.81007400	2.77996000
H	-2.63697100	-0.85511900	2.31249800
C	-1.54029000	-1.08387300	4.14298200
H	-2.42498400	-1.33804200	4.72001000
C	-0.29342500	-1.02910800	4.76343900
H	-0.20216300	-1.23900900	5.82508000
C	0.83735900	-0.70433300	4.01599600
H	1.81234900	-0.66166300	4.49292900
C	0.71821900	-0.43580300	2.65244800
H	1.61676500	-0.18839200	2.08945500
H	4.22297000	-2.31431800	-1.23561500
H	2.06764100	1.27277300	-0.17351200
H	-2.50593700	-2.54699800	-0.84528300
H	-3.38882500	1.65430000	-0.47472700

Singly-bridged SnPh₂-dimer (20) [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

Zero-point correction=	0.365021		
Thermal correction to Energy=	0.389985		
Thermal correction to Enthalpy=	0.390929		
Thermal correction to Gibbs Free Energy=	0.304843		
Sum of electronic and zero-point Energies=	-932.706577		
Sum of electronic and thermal Energies=	-932.681613		
Sum of electronic and thermal Enthalpies=	-932.680668		
Sum of electronic and thermal Free Energies=	-932.7766755		
Sn	-1.09848800	-2.21172500	-0.52193400
Sn	0.52550100	0.11169400	-0.90763300
C	2.67798300	0.11192200	-0.78509800
C	3.45708800	-0.52849700	-1.75734500
C	3.33189000	0.68420300	0.31670700
C	4.84682200	-0.58942600	-1.64123200
H	2.98337500	-0.99644900	-2.61830500
C	4.71765300	0.61907100	0.44212900
H	2.75570200	1.17445000	1.09878600
C	5.47790800	-0.01609600	-0.53987700
H	5.43364300	-1.08824800	-2.40712000
H	5.20385200	1.06152700	1.30688300
H	6.55838200	-0.06768100	-0.44334000
C	0.16033600	-1.28417800	1.20788400
C	1.35441800	-1.97430400	1.51525500
C	-0.25648000	-0.29648900	2.12261300
C	2.07476000	-1.71384900	2.67399700
H	1.73238800	-2.72574200	0.82391900
C	0.46513100	-0.02279800	3.28508800
H	-1.15700300	0.27803300	1.92401200
C	1.62966400	-0.73300100	3.56271500
H	2.98987900	-2.26097500	2.87933900
H	0.12014700	0.74999300	3.96550100
H	2.19441400	-0.52046700	4.46574400

C	-2.83472200	-0.86346400	-0.25187700
C	-3.40257000	-0.19633800	-1.34856100
C	-3.45994700	-0.69805600	0.99335600
C	-4.51834000	0.62813800	-1.20748200
H	-2.95486400	-0.29793400	-2.33598600
C	-4.57436700	0.12811200	1.14735100
H	-3.06916900	-1.20802800	1.87221500
C	-5.10269800	0.80016700	0.04655500
H	-4.92614000	1.14216300	-2.07330000
H	-5.03005200	0.24713400	2.12659200
H	-5.96658800	1.44778000	0.16360700
C	0.00443200	2.11814700	-0.30584700
C	-1.31093400	2.45525400	0.05125600
C	0.96669700	3.13811000	-0.32566200
C	-1.64657800	3.76260100	0.39702200
H	-2.08953700	1.69802700	0.06278000
C	0.63210100	4.44932900	0.01442400
H	1.99423700	2.91747000	-0.60449700
C	-0.67515600	4.76249000	0.37997900
H	-2.67061300	3.99759900	0.67259900
H	1.39333200	5.22377100	-0.00602500
H	-0.93660100	5.78286600	0.64487800

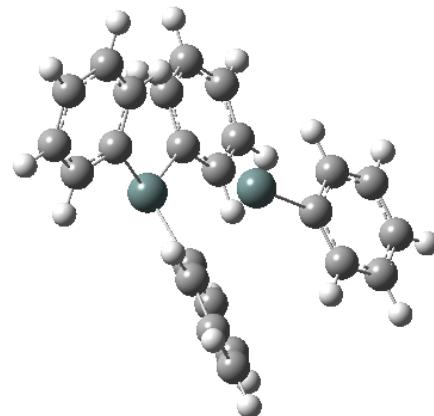
Doubly-bridged SnPh₂-dimer (*trans*-21) [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

Zero-point correction=	0.365170		
Thermal correction to Energy=	0.389986		
Thermal correction to Enthalpy=	0.390930		
Thermal correction to Gibbs Free Energy=	0.305331		
Sum of electronic and zero-point Energies=	-932.690084		
Sum of electronic and thermal Energies=	-932.665268		
Sum of electronic and thermal Enthalpies=	-932.664324		
Sum of electronic and thermal Free Energies=	-932.749923		
Sn	-0.95482800	0.00010300	-1.30369500
Sn	0.95501300	-0.00003300	1.30380300
C	0.00008100	1.79220000	0.00020900
C	-0.82613300	2.53658900	0.87165500
C	0.82632000	2.53675900	-0.87107000
C	-0.83414200	3.92831100	0.87244900
H	-1.47541900	2.01031400	1.56717100
C	0.83431700	3.92848300	-0.87160400
H	1.47564800	2.01062100	-1.56665600
C	0.00008000	4.62692000	0.00048300
H	-1.48327900	4.46764200	1.55571200
H	1.48346400	4.46795100	-1.55475100
H	0.00007700	5.71297800	0.00059100
C	-0.00007500	-1.79212700	0.00007100
C	0.82614900	-2.53669400	-0.87121600
C	-0.82633200	-2.53650900	0.87148300

C	0.83410100	-3.92841900	-0.87177900
H	1.47549900	-2.01056200	-1.56678400
C	-0.83438300	-3.92823000	0.87225000
H	-1.47562600	-2.01022800	1.56698900
C	-0.00016700	-4.62684700	0.00028500
H	1.48323800	-4.46789300	-1.55493000
H	-1.48355100	-4.46755500	1.55549000
H	-0.00020100	-5.71290500	0.00037200
C	-2.92900700	0.00009600	-0.30199700
C	-3.63779200	1.19612500	-0.10522400
C	-3.63742800	-1.19606400	-0.10470400
C	-4.96950100	1.20232800	0.31069900
H	-3.15178900	2.15431600	-0.27781900
C	-4.96913200	-1.20249400	0.31122700
H	-3.15112400	-2.15418300	-0.27685700
C	-5.63824100	-0.00013700	0.52906500
H	-5.48398600	2.14752000	0.46115200
H	-5.48332600	-2.14777700	0.46210400
H	-6.67412900	-0.00022500	0.85540300
C	2.92897100	-0.00009500	0.30178600
C	3.63746900	-1.19621500	0.10455000
C	3.63747200	1.19599500	0.10438400
C	4.96906300	-1.20255800	-0.31173500
H	3.15132100	-2.15436000	0.27701900
C	4.96906700	1.20227500	-0.31190300
H	3.15133300	2.15416800	0.27672300
C	5.63793700	-0.00015800	-0.53007400
H	5.48334200	-2.14780800	-0.46252700
H	5.48334900	2.14750200	-0.46282900
H	6.67372600	-0.00018300	-0.85672800

Transition State 25[‡] (17b ⇔ trans-21) [ωB97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

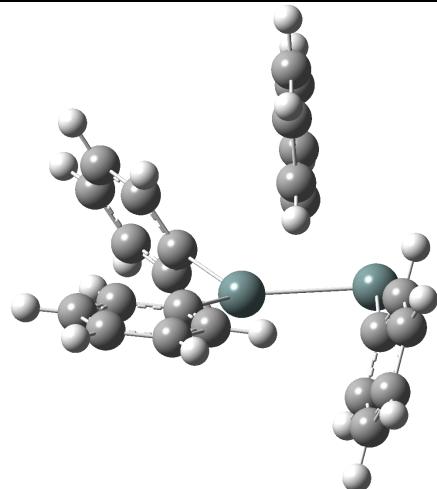
Zero-point correction=	0.364410
Thermal correction to Energy=	0.388593
Thermal correction to Enthalpy=	0.389537
Thermal correction to Gibbs Free Energy=	0.305896
Sum of electronic and zero-point Energies=	-932.659204
Sum of electronic and thermal Energies=	-932.635021
Sum of electronic and thermal Enthalpies=	-932.634077
Sum of electronic and thermal Free Energies=	-932.717718
Sn	0.23355100 -0.82735800 1.41369800
C	0.01199900 -1.51582200 -1.12270700
C	1.21307500 -1.66831200 -1.86265100
C	1.81759300 -2.90169800 -2.00101800
C	1.25819700 -4.04943700 -1.41070600
C	0.09646400 -3.93447000 -0.67173200
C	-0.52026300 -2.67900400 -0.50953100
H	2.73771800 -2.98272500 -2.57200200
H	-0.35581900 -4.80869700 -0.21354400
H	-1.48357200 -2.63641700 -0.00441300
H	1.66750400 -0.80458600 -2.33779800
H	1.73622800 -5.01536300 -1.53754300
Sn	-1.06711900 0.34827300 -1.11528700
C	0.05040800 2.15917500 -0.85331000
C	-0.62473600 3.31517100 -0.43488400
C	0.04463200 4.53276000 -0.32839900
C	1.39834700 4.61998700 -0.65052900
C	2.07490400 3.48049400 -1.08012400
C	1.40757000 2.26169400 -1.18276000
H	-0.49457600 5.41516300 0.00386800
H	3.13123500 3.53301600 -1.32742500
H	1.96944400 1.38590000 -1.49363800
H	-1.68388000 3.27626800 -0.19089500
H	1.91927800 5.56906100 -0.56977400
C	-2.90885400 0.17703800 -0.01220300
C	-3.21872000 0.95889600 1.11037500
C	-3.86613900 -0.74273600 -0.46160900
C	-4.44452600 0.82626800 1.75911100
H	-2.49210400 1.66818800 1.49888700
C	-5.08472000 -0.89460400 0.20078900
H	-3.66920400 -1.34942000 -1.34280600
C	-5.37931500 -0.10608700 1.30984100
H	-4.66435700 1.44250800 2.62621800
H	-5.80742400 -1.62136500 -0.15871100
H	-6.33074200 -0.21586600 1.82119700
C	2.35707900 -0.39249000 1.09440200
C	2.81682500 0.90701600 1.35657500
C	3.30954100 -1.35785900 0.73369800
C	4.16716100 1.23862100 1.24111800
H	2.11520000 1.69402200 1.63109600



C	4.66007700	-1.03475100	0.61135500
H	3.00005900	-2.38189300	0.53596500
C	5.09174000	0.26799500	0.86219400
H	4.49268500	2.25652200	1.43712900
H	5.37716800	-1.79875400	0.32358100
H	6.14350700	0.52159400	0.76630300

Transition State 26[‡] [ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output]

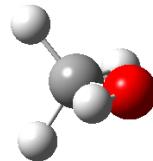
Zero-point correction=	0.364501
Thermal correction to Energy=	0.388796
Thermal correction to Enthalpy=	0.389740
Thermal correction to Gibbs Free Energy=	0.304982
Sum of electronic and zero-point Energies=	-932.706224
Sum of electronic and thermal Energies=	-932.681929
Sum of electronic and thermal Enthalpies=	-932.680985
Sum of electronic and thermal Free Energies=	-932.765743
Sn	-1.03747300
Sn	0.60671900
C	2.75964900
C	3.50707100
C	3.45231400
C	4.90125400
H	3.00407000
C	4.84322400
H	2.90312500
C	5.57063200
H	5.46159800
H	5.35961900
H	6.65478000
C	0.19588900
C	1.25277800
C	-0.51046400
C	1.57730200
H	1.83827100
C	-0.19121800
H	-1.31699700
C	0.85141900
H	2.39381000
H	-0.74877000
H	1.10085400
C	-2.79003800
C	-3.23845900
C	-3.54818000
C	-4.37430300
H	-2.68055100
C	-4.67790100
H	-3.25056100
C	-5.09170300
	-2.38189300
	0.26799500
	2.25652200
	-1.79875400
	0.52159400
	0.86219400
	1.43712900
	0.32358100
	0.76630300



H	-4.68921500	1.49783700	-2.25365000
H	-5.23533600	0.02439400	1.74463300
H	-5.96960600	1.48987200	-0.12355300
C	0.09003800	2.14488500	-0.09453500
C	-1.23174300	2.48831300	0.22836700
C	1.07812100	3.13252700	0.01997800
C	-1.55126100	3.77216900	0.66618100
H	-2.02846900	1.75532200	0.13835700
C	0.76018100	4.42022200	0.45324100
H	2.11268200	2.90302100	-0.22506300
C	-0.55568400	4.74127000	0.77964300
H	-2.58150300	4.01361900	0.91174900
H	1.54082200	5.17105000	0.53553000
H	-0.80442600	5.74312700	1.11745900

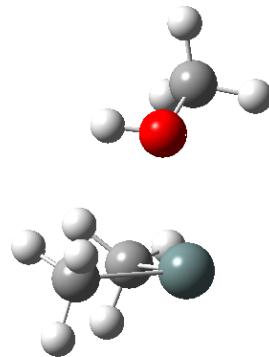
Methanol (ω B97XD/6-31+G(d,p) output)

Zero-point correction=	0.051874
Thermal correction to Energy=	0.055216
Thermal correction to Enthalpy=	0.056160
Thermal correction to Gibbs Free Energy=	0.029110
Sum of electronic and zero-point Energies=	-115.643298
Sum of electronic and thermal Energies=	-115.639957
Sum of electronic and thermal Enthalpies=	-115.639013
Sum of electronic and thermal Free Energies=	-115.666062
C	0.66276400 -0.02069000 0.00000300
H	1.08189800 0.98703500 -0.00054600
H	1.02462200 -0.54512500 0.89426900
O	-0.74485300 0.12210500 0.00000300
H	-1.14881200 -0.74858300 0.00000000
H	1.02453300 -0.54603100 -0.89376700

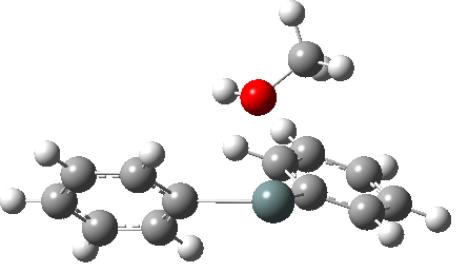


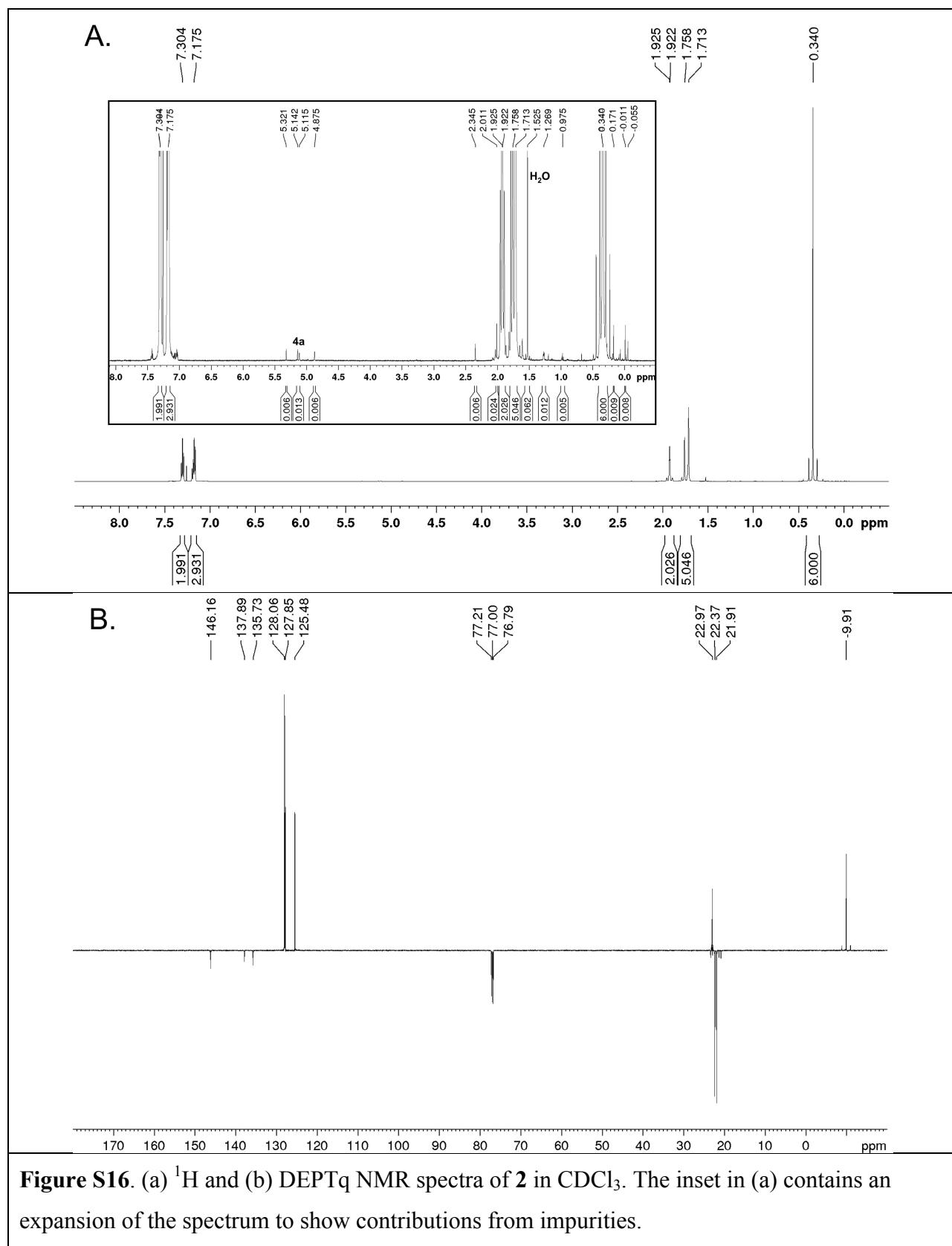
SnMe₂-MeOH Complex (18) (ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output)

Zero-point correction=	0.124995
Thermal correction to Energy=	0.135156
Thermal correction to Enthalpy=	0.136100
Thermal correction to Gibbs Free Energy=	0.088812
Sum of electronic and zero-point Energies=	-198.754438
Sum of electronic and thermal Energies=	-198.744276
Sum of electronic and thermal Enthalpies=	-198.743332
Sum of electronic and thermal Free Energies=	-198.790620
Sn	-0.54634400 0.11873500 -0.44124500
C	2.76060200 -0.05030700 -0.02260000
H	2.75638200 0.36790500 -1.02976800
H	2.73696500 0.76392500 0.70774100
H	3.65796800 -0.65970700 0.11255900
O	1.58900600 -0.86417500 0.08044400
C	-1.28861000 -1.38119900 1.00117100
H	-1.18784100 -2.40282800 0.61793600
H	-0.78341900 -1.30965400 1.97346100
H	-2.35708000 -1.21010900 1.18414700
C	-0.08374900 1.60818200 1.12835900
H	0.53585000 2.43592400 0.76592600
H	-1.02396900 2.04152800 1.49455200
H	0.41185200 1.14480700 1.99147500
H	1.52898200 -1.25521900 0.95909000



SnPh₂-MeOH Complex (19) (ω B97XD/6-31+G(d,p)-LANL2DZdp(Sn) output)

Zero-point correction=	0.236251			
Thermal correction to Energy=	0.252077			
Thermal correction to Enthalpy=	0.253021			
Thermal correction to Gibbs Free Energy=	0.190421			
Sum of electronic and zero-point Energies=	-581.998507			
Sum of electronic and thermal Energies=	-581.982681			
Sum of electronic and thermal Enthalpies=	-581.981737			
Sum of electronic and thermal Free Energies=	-582.044337			
C	1.56336900	0.28661900	-0.30611700	
C	-1.71471000	0.15098100	-0.29330700	
C	1.42900300	1.31775100	0.63923000	
C	2.49754400	2.14652700	0.97917500	
C	3.73865500	1.96885000	0.36704600	
C	3.90061700	0.95920900	-0.57872800	
C	2.82486100	0.12842200	-0.90059200	
C	-1.71618600	1.52604500	-0.58048400	
C	-2.88373800	-0.38584400	0.27257400	
C	-2.82488700	2.32843300	-0.30951300	
C	-3.96484900	1.77269100	0.26971000	
C	-3.99299600	0.41066300	0.56285400	
H	2.36283100	2.93413900	1.71590700	
H	4.86356800	0.81506800	-1.06094700	
H	2.98172600	-0.66262300	-1.63336500	
H	-2.94178300	-1.45229400	0.48996900	
H	-2.79680400	3.38907300	-0.54425100	
H	0.46592100	1.49104000	1.11788700	
H	4.57227000	2.61521100	0.62612300	
H	-0.83315200	1.98810300	-1.01766700	
H	-4.82724100	2.39625700	0.48690500	
H	-4.88013300	-0.03267300	1.00686200	
Sn	-0.02053400	-1.18294000	-0.76397300	
C	1.06656700	-1.84364300	2.33414300	
H	1.49506200	-2.73489700	1.87458100	
H	1.78274700	-1.01952500	2.26688000	
H	0.81788900	-2.05707500	3.37672900	
O	-0.12549200	-1.54385600	1.59847500	
H	-0.57179200	-0.77220200	1.96929100	



References

- (1) Carmichael, I.; Helman, W. P.; Hug, G. L. *J. Phys. Chem. Ref. Data* **1987**, *16*, 239.
- (2) Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Scalmani, G.; Barone, V.; Mennucci, B.; Petersson, G. A.; Nakatsuji, H.; Caricato, M.; Li, X.; Hratchian, H. P.; Izmaylov, A. F.; Bloino, J.; Zheng, G.; Sonnenberg, J. L.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Vreven, T.; Montgomery, J., J. A.; Peralta, J. E.; Ogliaro, F.; Bearpark, M.; Heyd, J. J.; Brothers, E.; Kudin, K. N.; Staroverov, V. N.; Kobayashi, R.; Normand, J.; Raghavachari, K.; Rendell, A.; Burant, J. C.; Iyengar, S. S.; Tomasi, J.; Cossi, M.; Rega, N.; Millam, N. J.; Klene, M.; Knox, J. E.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Martin, R. L.; Morokuma, K.; Zakrzewski, V. G.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Dapprich, S.; Daniels, A. D.; Farkas, Ö.; Foresman, J. B.; Ortiz, J. V.; Cioslowski, J.; Fox, D. J.; Gaussian, I., Wallingford CT, 2009; Gaussian, Inc.: Wallingford CT, 2009.